

# (12) UK Patent Application (19) GB (11) 2 236 325 (13) A

(43) Date of A publication 03.04.1991

(21) Application No 9019046.3

(22) Date of filing 31.08.1990

(30) Priority data

(31) 01223079

(32) 31.08.1989

(33) JP

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(51) INT CL<sup>5</sup>

C22C 45/00

(52) UK CL (Edition K)

C7A AA23X AA23Y AA230 AA231 AA233 AA235  
AA237 AA239 AA24X AA241 AA243 AA245 AA247  
AA249 AA25X AA25Y AA250 AA253 AA255 AA257  
AA259 AA260 AA263 AA266 AA269 AA27X AA272  
AA276 AA279 AA28X AA28Y AA280 AA289 AA290  
AA293 AA296 AA299 AA30X AA30Y AA300 AA303  
AA305 AA307 AA309 AA31X AA311 AA313 AA316  
AA319 AA320 AA323 AA326 AA329 AA33X AA33Y  
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AA557 AA559 AA56X AA562 AA565 AA568 AA57Y  
AA571 AA574 AA577 AA579 AA58X AA58Y AA584  
AA587 AA589 AA59X AA591 AA593 AA595 AA599

((52) (56) and (58) continued overleaf)

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(54) Thin-aluminum-based alloy foil and wire

(57) A thin aluminum-based alloy foil or wire is produced from an amorphous material made by a quenching and solidifying process and having a composition represented by the general formula:



wherein M is one or more elements selected from a group consisting of V, Cr, Mn, Fe, Co, Ni, Cu, Zr, Ti, Mo, W, Ca, Li, Mg and Si; X is one or more elements selected from a group consisting of Y, Nb, Hf, Ta, La, Ce, Sm, Nd and Md (misch metal); each of a, b, and c are an atomic percentage, with the proviso that

$$\begin{aligned} 50 &\leq a \leq 95 \\ 0.5 &\leq b \leq 35 \text{ and} \\ 0.5 &\leq c \leq 25 \end{aligned}$$

Such foil or wire has a smooth surface and a very small and uniform foil thickness or wire diameter, contains at least 50% by volume of an amorphous phase, and has excellent strength and resistance to corrosion. The foil thickness and wire diameter are reduced in a rolling or drawing process at an elevated temperature over a short time period.

GB 2 236 325 A

(52) UK CI (Edition K) continued

AA609 AA61X AA61Y AA613 AA615 AA617 AA619  
AA62X AA621 AA623 AA625 AA627 AA629 AA67X  
AA671 AA673 AA674 AA675 AA677 AA679 AA68X  
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AB636 AB66X AB661 AB663 AB665 AB667 AB669  
AB670 AB675 AB70X AB702 A80X  
U1S S1570 S1574

(56) Documents cited  
None

(58) Field of search  
UK CL (Edition K) C7A  
INT CL<sup>6</sup> C22C

FIG.1

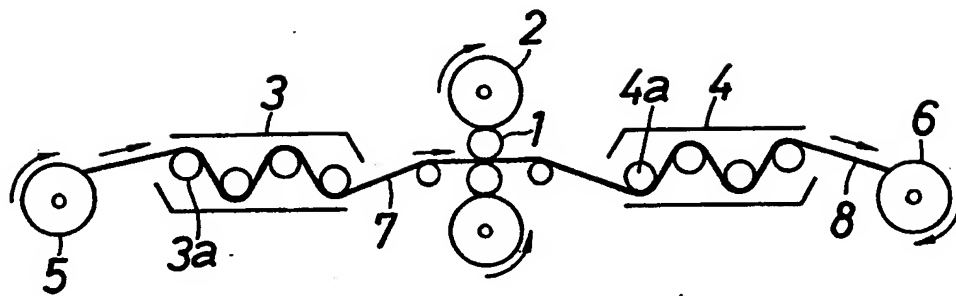
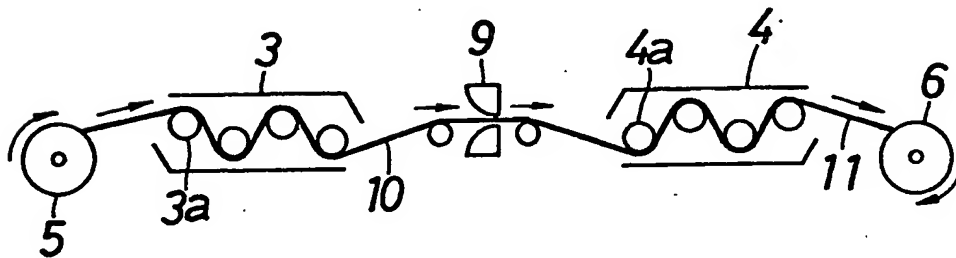


FIG.2



## THIN ALUMINUM-BASED ALLOY FOIL AND WIRE

The field of the present invention is thin aluminum-based alloy foils and wires which are excellent in strength and corrosion resistance, have a smooth surface, and have a very small thickness or diameter with a uniform distribution of thickness or diameter thereof, and a process for producing same.

The present inventors have already developed alloys within a wider range of compositions based on aluminum and have filed patent applications therefore, such as Japanese Patent Applications Laid-Open Nos. 47831/89; 127641/89; 240632/89; 240631/89; and 275732/89.

Such alloys are being studied for application to wider fields of structural members for vehicles, corrosion-resistant materials for chemical apparatus, corrosion- or wear-resistant coating materials and the like as materials exhibiting excellent specific strength (strength/alloy density), corrosion resistance and stability in high temperature, and workability.

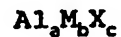
Conventional amorphous alloys have been produced in the form of a ribbon, a wire, a powder or a coating film by a

liquid quenching process, a submerged spinning process, a gas-atomizing process, or a physical or chemical vapor deposition process. In such case, it is difficult to produce an amorphous ribbon of a thickness of 10  $\mu\text{m}$  or less and an amorphous wire of a diameter of 50  $\mu\text{m}$  or less. In addition, the materials such as the amorphous ribbon, wire or the like are non-uniform in thickness or diameter and also have a greater surface roughness. For this reason, such materials cannot be directly utilized in fields of applications in which an extremely small thickness, an extremely small fineness, a smoothness in surface and a uniformity in thickness and in diameter are required. Moreover, such materials are higher in hardness and strength, and currently it is impossible to easily effect the usual working processes such as rolling or drawing of such materials which otherwise might be effective for overcoming the above disadvantages.

It is an object of the present invention to provide an aluminum-based alloy foil or a thin aluminum-based alloy wire having a smooth surface and a uniform thickness or diameter while substantially maintaining the desirable properties, such as strength, possessed by an amorphous alloy ribbon or wire.

To achieve the above object, according to the present invention, there is provided an aluminum-based alloy foil or a thin aluminum-based alloy wire having excellent strength and

resistance to corrosion, which is produced from a material made by quenching and solidifying process and having a composition represented by the general formula:



wherein M is one or more elements selected from the group consisting of V, Cr, Mn, Fe, Co, Ni, Cu, Zr, Ti, Mo, W, Ca, Li, Mg and Si; X is one or more elements selected from the group consisting of Y, Nb, Hf, Ta, La, Ce, Sm, Nd and Md (misch metal); each of a, b and c is an atomic percentage, wherein

$$50 \leq a \leq 95$$

$$0.5 \leq b \leq 35 \text{ and}$$

$$0.5 \leq c \leq 25,$$

and which has a substantially smooth surface and a very small and substantially uniform thickness or diameter and contains at least 50% by volume of an amorphous phase. In addition, there is also provided a process for producing an aluminum-based alloy foil or a thin aluminum-based alloy wire of the type described above, comprising rolling or drawing an amorphous material having a composition represented by the above general formula at a working temperature within a glass transition region, super-cooled liquid region or crystallization starting temperature  $\pm 100^\circ\text{K}$  that is specific to the amorphous material.

The aluminum-based alloy foil according to the present invention is an alloy foil which is very thin and has a

beautiful surface and a uniform thickness, as well as excellent strength, hardness and resistance to corrosion, and thus, it is useful as a laminate material requiring a corrosion-resistant property such as in food and chemical fields, or as a magnetic recording metal tape substrate, or as a brazing material for precision machinery. In addition, the thin aluminum-based alloy wire according to the present invention is an extremely thin alloy wire having excellent strength and resistance to corrosion and thus, it is useful as a filler for composite materials such as concretes, metals and resins.

Further, with the process according to the present invention, it is possible to efficiently produce an aluminum-based alloy foil or a thin aluminum-based alloy wire having excellent properties described above.

The above and other objects, features and advantages of the invention will become apparent from a reading of the following description of the preferred embodiment taken in conjunction with the accompanying drawings, in which:

Fig. 1 is a diagram illustrating a rolling machine for producing an amorphous alloy foil and

Fig. 2 is a diagram illustrating a drawing machine for producing a thin amorphous alloy wire.

Using various aluminum alloys representative of Al-Ni-Y based alloys, for example, as described in Japanese Patent Application Laid-Open No. 47831/89, amorphous alloy ribbons having a width of 1 to 300 mm and a thickness of 5 to 500  $\mu\text{m}$  or amorphous alloy wires having a diameter of 0.01 to 1 mm can be produced by the utilization of a quenching and solidifying process. However, it is difficult to produce a high quality alloy foil or fine wire having a thickness of 10  $\mu\text{m}$  or less or a diameter of 50  $\mu\text{m}$  or less by such process. If such a member is intended to be produced, the resulting product may be of a partially non-uniform thickness or diameter and sometimes may have defects such as pores produced therein. Therefore, it is difficult to stably and continuously produce a high quality ribbon or wire. In order to stably and continuously produce a high quality ribbon or wire, the thickness of the ribbon has been limited to a range of 15 to 100  $\mu\text{m}$ , while the diameter of the wire has been limited to a range of 80 to 150  $\mu\text{m}$ .

The amorphous alloys show various glass transition temperatures  $T_g$  and crystallization temperatures  $T_x$  in an alloy composition within a range represented by the above-described general formula. In a region of temperatures between  $T_x$ - $T_g$ , the alloys have the characteristic of a supercooled liquid while it is of a solid phase, and easily exhibit large plastic deformations under a very low stress. Some of such large plastic deformations reach 500% by applying simple tension (by



loading of a uniaxial stress). Near to the crystallization temperature ( $T_x \pm 100^\circ\text{K}$ ), the alloys generate a super plasticity phenomenon and likewise exhibit large plastic deformations under a very low stress.

By paying attention to these characteristics and by selecting a rolling or drawing temperature within the glass transition region or supercooled liquid region, or near to the crystallization temperature, a rolling or drawing can be easily conducted to provide an aluminum-based alloy foil or a fine aluminum-based alloy wire including at least 50% by volume of an amorphous phase and having a foil thickness of 10  $\mu\text{m}$  or less or a wire diameter of 50  $\mu\text{m}$  or less.

Here, the term "crystallization temperature  $T_x$ " means a starting temperature ( $^\circ\text{K}$ ) of an exothermic peak initially appearing in a differential scanning calorimetric profile provided by heating an amorphous material under ambient pressure at a heating rate of  $40^\circ\text{K}/\text{min}$ , and the term "glass transition temperature  $T_g$ " indicates a starting temperature ( $^\circ\text{K}$ ) of an endothermic peak initially appearing near a point below the crystallization temperature  $T_x$ .

It is commonly known that an amorphous alloy exhibits a large plastic deformation even at ambient temperature under a multi-axial stress, but the advantages of the process according to the present invention are in that working can be effected under a lower than normal stress and a higher rolling reduction

(rate of reduction in section) of 50% or more and further that even a relatively brittle material that is difficult to roll or draw at ambient temperature can be easily worked. That is, it is possible to easily produce a continuous foil or a thin wire having a foil thickness of 10  $\mu\text{m}$  or less or a wire diameter of 50  $\mu\text{m}$  or less by the process of rolling or drawing, at one or two stages, a ribbon of a thickness of about 15 to 100  $\mu\text{m}$  or a wire of a diameter of about 80 to 150  $\mu\text{m}$ , which ribbon or wire is of an alloy composition within the above-described range and produced by a usual liquid quenching process.

The foils or thin wires produced by such process not only have a smooth surface and a uniform thickness or diameter, but also maintain the amorphous property of the amorphous ribbon or the like and exhibit excellent strength and resistance to corrosion. Some of such foils or thin wires may exhibit an increase in strength of 10 to 20% and an increase in ductility of 5 to 20% depending upon the alloy composition.

The stage of crystallization of an amorphous material proceeds with a balance of the temperature of the material and the time of retention thereof. If the temperature of the material is lower than the crystallization temperature  $T_x$ , the material is crystallized in a shorter time at a temperature nearer to the crystallization temperature  $T_x$ . If the temperature of the material is higher than the crystallization temperature  $T_x$ , the material is crystallized at a shorter time

at a temperature farther from the crystallization temperature  $T_x$ .

In order to produce an alloy foil or thin alloy wire including at least 50% by volume of an amorphous phase by rolling or drawing an amorphous ribbon or wire having the above-described alloy composition according to the present invention, it is desirable that the working temperature is determined in a range approximately equal to the crystallization temperature  $T_x \pm 100^\circ\text{K}$ , preferably the crystallization temperature  $T_x \pm 30^\circ\text{K}$ , more preferably the crystallization temperature  $T_x - 30^\circ\text{K}$ , and that the working including all the heating, working and cooling steps is completed within 150 sec.

With amorphous materials having a composition as represented by the above-described general formula, however, most of them show a wider over-cooled liquid region  $T_x-T_g$  and within this region, the time of crystallization is largely delayed and hence wider acceptable ranges of working temperature and time can be employed.

More specifically, the aluminum alloy-based amorphous material having the alloy composition according to the present invention has a supercooled liquid region  $T_x-T_g$  in a range of 10 to  $20^\circ\text{K}$ , and, therefore, an alloy foil or a thin alloy wire including at least 50% by volume of an amorphous phase can be produced from this amorphous material even by setting the rolling or drawing temperature in this temperature region and

using a working time within 600 sec. The working time is not independent and is determined depending upon the working temperature used and hence, the working time can be more prolonged by employing a lower working temperature.

As described above, in order to produce an alloy foil or a thin alloy wire comprising an amorphous phase, it is desirable that the entire working process including heating, working and cooling steps is completed within a time of 150 sec to 600 sec, depending on the material. For this purpose, it is essential to heat the material to the working temperature in a short time immediately before rolling or drawing and to cool the material immediately after working to a temperature ( $T_x - 200^\circ\text{K}$  or less is preferred) at which the amorphous phase will not be phase-converted to a crystalline phase.

The actual working is conducted by a procedure which will be described below with reference to the drawings.

In producing an amorphous alloy foil, as shown in Fig. 1, a heating device 3 is disposed immediately upstream of work rolls 1 of a rolling machine and includes a plurality of heating rolls 3a. The heating rolls 3a are heated by an electro-thermic source or any other conventional heat source and their temperature is controllable. In addition, a cooling device 4 is disposed immediately downstream of the work rolls 1 and includes a plurality of cooling rolls 4a which are cooled by water or another cooling medium. Thus, an

amorphous ribbon 7 supplied from an unwinder 5 is heated to a predetermined working temperature through the heating device 3 while being continuously brought into contact with the individual heating rolls 3a and then, the heated ribbon is immediately rolled to a predetermined thickness by the work rolls 1. Subsequently, the amorphous alloy foil 8 produced by the rolling is immediately cooled to a predetermined temperature through the cooling device 4 while being continuously brought into contact with the individual cooling rolls 4a and is then taken up by a winder 6. The work rolls 1 are each supported by a back-up roll 2.

Fig. 2 illustrates a drawing machine for producing a fine amorphous alloy wire, wherein reference numeral 9 identifies a drawing die; reference numeral 10 identifies an amorphous wire; and reference numeral 11 identifies a fine amorphous alloy wire. The other components are the same as in Fig. 1 and hence, are designated by the same reference characters and the description thereof is omitted. In this case, a heating means also can be included in the drawing die 9.

The pluralities of heating and cooling rolls 3a and 4a within the heating and cooling devices 3 and 4 are rotated synchronously with a travel speed of the amorphous ribbon 7, amorphous wire 10, or the like.

By using the heating rolls 3a and the cooling rolls 4a as described above, the amorphous ribbon 7, amorphous wire 10, or the like can be rapidly heated and the amorphous alloy foil 8,

amorphous wire 10, or the like can be rapidly cooled. It is also possible to use various other means for heating, such as by radiation from an electric heater or a heating box through which a high temperature gas convects, or a means for heating by contact of a high speed and high temperature gas with the amorphous ribbon 7, amorphous wire 10, or the like. Various other means for cooling may be used, such as, by contact with water or a high speed and low temperature gas by the amorphous alloy foil 8, amorphous wire 10, or the like. When the working speed is reduced, the amorphous ribbon 7 may be heated concurrently with rolling by including a heating device in the work roll 1 without provision of the heating device 3.

For purposes of further description without limiting the scope of the invention, specific examples of the product and process of this invention will now be described in further detail. Amorphous alloy foils 8 were produced using the rolling machine shown in Fig. 1. The starting materials prepared were five types of amorphous ribbons 7 coiled and having alloy compositions given in Table I with a thickness of 20  $\mu\text{m}$  and a width of about 20 mm.

The heating device 3 was disposed at a place 30 cm upstream of the work rolls 1, and the cooling device 4 was disposed at a place 30 cm downstream of the work rolls 1. The heating device 3 included four heating rolls 3a of a diameter of 60 mm, each of which was controlled in temperature by an electric heating, while the cooling device 4 included

four cooling rolls 4a of a diameter of 60 mm, each of which was cooled by water.

The work rolls 1 used were of a diameter of 20 mm, and heating of each work roll 1 was provided by conduction from the back-up roll 2. In this case, the heating temperature of the back-up roll 2 was set at near the desired working temperature for each amorphous ribbon 7.

The rolling temperature was set at a level of equal to the crystallization temperature  $T_x$  of each ribbon 7, minus  $30^\circ\text{K}$ , within  $\pm 5^\circ\text{K}$  or at a level equal to the temperature at the central portion of the supercooled liquid region of each ribbon 7, within  $\pm 5^\circ\text{K}$ . The rolling rate was set at 20 m/min, and the rearward tension on the amorphous ribbon 7 was set at 20 kg.

The following steps were continuously conducted, as generally described above, the step of providing an amorphous ribbon 7 around the unwinder 5, the step of passing the amorphous ribbon 7 as it is unwound from the unwinder 5 through the heating device 3 to heat it to the working temperature, the step of subjecting the amorphous ribbon 7 to the rolling to produce an amorphous alloy foil 8, the step of passing the amorphous alloy foil 8 through the cooling device 4 to cool it to approximately room temperature, and the step of taking up the amorphous alloy foil 8 around the winder 6.

Each amorphous alloy foil 8 thus produced was of a thickness of about  $7\text{ }\mu\text{m}$  and a width of about 20 mm and had a

beautiful surface and a uniform thickness with inaccuracies of  $\pm 0.1 \mu\text{m}$  or less both across the width and along the length of the foil 8.

Each foil 8 was examined for its structure by an X-ray diffraction and measured for tensile strength to provide the results given in Table I. In Table I, Amo means that the amorphous phase is of 100%; St. means Structure; Thi. means Thickness; Wid. means Width; and Stre. means Strength.

As apparent from Table I, it was ascertained that all the foils 8 were of an amorphous phase and had extremely excellent mechanical properties with a tensile strength of 1050 MPa or more.

TABLE I

Alloy composition (atomic % in the subscripts)	Ribbon		Foil			
	Tg (°K)	Tx (°K)	St. —	Thi. ( $\mu\text{m}$ )	Wid. (mm)	Stre. (Mpa)
$\text{Al}_{80}\text{Fe}_{10}\text{Nb}_{10}$	—	753	Amo	6.5	20	1050
$\text{Al}_{80}\text{Co}_{10}\text{Nb}_{10}$	—	697	Amo	7.2	20	1125
$\text{Al}_{85}\text{Ni}_5\text{Y}_{10}$	535	560	Amo	7.0	20	1210
$\text{Al}_{85}\text{Cu}_{10}\text{Mn}_5$	538	552	Amo	6.8	20	1120
$\text{Al}_{80}\text{Ni}_5\text{Fe}_5\text{Ce}_{10}$	615	633	Amo	7.0	20	1050

As further, examples of the product and process of this invention, thin amorphous alloy wires 11 were produced using a drawing machine as shown in Fig. 2.

Starting materials prepared were coils of four types of amorphous wires 10 of a diameter of  $100 \mu\text{m}$  and having the alloy compositions given in Table II.



The heating device 3 was disposed at a place 30 cm immediately upstream of the drawing dies 9, and the cooling device 4 was disposed at a place 30 cm immediately downstream of the drawing dies 9. The heating device 3 included four heating rolls 3a of a diameter of 60 mm, each of which was controlled in temperature by an electric heater, while the cooling device 4 included four cooling rolls 4a of a diameter 60 mm, each of which was cooled by water.

The drawing dies 9 were heated by an electric heater. The heating temperature of the drawing dies 9 was set at near the desired working temperature of each amorphous wire 10.

The drawing temperature was set at a level equal to the crystallization temperature  $T_x$  of each amorphous wire 10, minus  $30^\circ\text{K}$ , within  $\pm 5^\circ\text{K}$  or at a level equal to the temperature at the central portion of a supercooled liquid region of each amorphous wire 10, within  $\pm 5^\circ\text{K}$ . The drawing rate was set at 5 m/min.

The following steps were continuously conducted, as generally described above: the step of providing the amorphous wire 10 around the unwinder 5, the step of passing the amorphous wire 10 as it is unwound from the unwinder 5 through the heating device 3 to heat it to the drawing temperature, the step of subjecting the amorphous wire 10 to the drawing to fabricate a thin amorphous alloy wire 11, the step of passing the thin amorphous alloy wire 11 through the cooling device 4 to cool it to approximately room temperature,

and the step of taking up the amorphous alloy wire 11 around the winder 6.

Each thin amorphous alloy wire 11 thus produced was of a diameter of about 8  $\mu\text{m}$  and had a beautiful surface and a uniform diameter with inaccuracies of  $\pm 0.1 \mu\text{m}$  along the length of the wire 11.

Each thin wire 11 was examined for its structure by an X-ray diffraction and measured for its tensile strength to provide the results given in Table II. In Table II, the various legends have the same meaning as those legends in Table I.

As apparent from Table II, it was ascertained that all the thin wires 11 were of an amorphous phase and had extremely excellent mechanical properties with a tensile strength of 980 MPa or more.

TABLE II

Alloy composition (Atomic in the Subscripts)	Wire		Thin Wire		
	Tg (°K)	Tx (°K)	Structure	Diameter ( $\mu\text{m}$ )	Strength (Mpa)
$\text{Al}_{85}\text{Co}_5\text{Ce}_{10}$	607	615	Amo	8	980
$\text{Al}_{78}\text{Cr}_3\text{Cu}_7\text{Ce}_{12}$	-	605	Amo	8	1060
$\text{Al}_{86}\text{Ni}_4\text{Y}_{10}$	525	535	Amo	8	1205
$\text{Al}_{75}\text{Ni}_8\text{Si}_2\text{Mn}_{15}$	639	654	Amo	8	1085

The foregoing examples of thin aluminum-based alloy foils and wires and the processes for making same are illustrative of the invention and are not intended to be exhaustive of the

products or processes within the scope of this invention as defined by the following claims.

CLAIMS

1. A thin aluminum-based alloy foil or wire having excellent strength and resistance to corrosion, which is produced from a material made by a quenching and solidifying process and having a composition represented by the general formula:



wherein M is one or more elements selected from a group consisting of V, Cr, Mn, Fe, Co, Ni, Cu, Zr, Ti, Mo, W, Ca, Li, Mg and Si; X is one or more elements selected from a group consisting of Y, Nb, Hf, Ta, La, Ce, Sm, Nd and Md (misch metal); each of a, b and c is an atomic percentage wherein

$$\begin{aligned} 50 &\leq a \leq 95 \\ 0.5 &\leq b \leq 35 \text{ and} \\ 0.5 &\leq c \leq 25, \end{aligned}$$

and which has a substantially smooth surface and a very small and substantially uniform thickness of the foil or diameter of the wire and contains at least 50% by volume of an amorphous phase.

2. A foil or wire as claimed in claim 1, which is formed from a starting ribbon thicker than said foil or a starting wire of a larger diameter than said wire, respectively, of said material by a process in which said material is mechanically worked to reduce the thickness of the starting ribbon and the diameter of the starting wire.

3. A foil or wire as claimed in claim 2 wherein the step in which said material is mechanically worked is carried out at an elevated temperature.

4. A foil or wire as claimed in claim 3, wherein said elevated temperature is a glass transition temperature, supercooled liquid region temperature or crystallization starting temperature  $\pm 100^\circ\text{K}$  that is specific to said material.

5. A foil or wire as claimed in any one of claims 2 to 4 wherein the starting ribbon thickness or the starting wire diameter are reduced by at least 50% by the mechanical working.

6. A wire as claimed in any one of claims 2 to 5 wherein the starting wire diameter is reduced by at least 90% by the mechanical working.

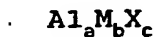
7. A foil as claimed in any one of claims 1 to 5 having a thickness of less than about  $10\text{ }\mu\text{m}$ .

8. A wire as claimed in any one of claims 1 to 6 having a diameter of less than  $50\text{ }\mu\text{m}$ .

9. A wire as claimed in any one of claims 1 to 6 and 8 having a diameter of less than  $10\text{ }\mu\text{m}$ .

10. A thin aluminum-based alloy foil or wire as claimed in claim 1 substantially as herein described.

11. A process for producing a thin aluminum-based alloy foil or wire having excellent strength and resistance to corrosion, comprising rolling or drawing an amorphous material made by a quenching and solidifying process and having a composition represented by the general formula:



wherein M is one or more elements selected from a group consisting of V, Cr, Mn, Fe, Co, Ni, Cu, Zr, Ti, Mo, W,

Ca, Li, Mg and Si; X is one or more elements selected from a group consisting of Y, Nb, Hf, Ta, La, Ce, Sm, Nd and Md (misch metal); each of a, b and c is an atomic percentage, wherein

$$\begin{aligned} 50 &\leq a \leq 95 \\ 0.5 &\leq b \leq 35 \text{ and} \\ 0.5 &\leq c \leq 25 \end{aligned}$$

at a working temperature within a glass transition region, supercooled liquid region or crystallization starting temperature  $\pm 100^\circ\text{K}$  that is specific to the amorphous material.

12. A process as claimed in claim 11, wherein said material is heated to said working temperature, the rolling or drawing is performed, and the material is cooled to approximately room temperature within a 600 sec. time period.

13. A process as claimed in claim 12, wherein said time period is less than 150 sec.

14. A process as claimed in claim 12 or claim 13 wherein said time period is determined by the characteristics of the amorphous material for maintaining an amorphous phase of at least 50%.

15. A process as claimed in any one of claims 11 to 14 wherein said working temperature is equal to the crystallization temperature  $T_x \pm 30^\circ\text{K}$  of the amorphous material.

16. A process as claimed in any one of claims 11 to 14 wherein said working temperature is within the range between the crystallization temperature of the amorphous material and  $30^\circ\text{K}$  less than said crystallization temperature.

17. A process as claimed in claim 11 substantially as herein described and with reference to the Examples.